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1.1 Molecular Mass Transfer For a binary mixture of A and B, the molar flux, $N_{A,z}$, of species A relative to the z axis is $N_{A,z} = -cD \frac{dy_A}{dz} + y_A(N_{A,z} + N_{B,z})$ (1.1) In this equation, c is the total molar concentration, D is the diffusivity of A in B, y_A is the mole fraction of A, and $N_{B,z}$ is the molar flux of B.

Chapter 1 Fundamentals of Mass Transfer

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mass transfer will be used throughout this module. 10.1 Mass transfer through diffusion . In Module 2 "Conduction", the Fourier equation was introduced, which relates the heat transfer to an existent temperature gradient: $q = -k \frac{dT}{dy}$ (Fourier's law) (10.1) For mass transfer, where a component A diffuses in a mixture with a component

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mass transfer operation because the fluid adheres to the solid surface instead of dissolving in the solid. Adsorption involves the transfer of mass from the solid surface (adsorbents) to the gas or liquid medium (adsorbates). A few operations of practical

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The subject of Mass Transfer studies the relative motion of some chemical species with respect to others (i.e. separation and mixing processes), driven by concentration gradients (really, an imbalance in chemical potential, as explained in). Fluid flow without mass transfer is not part of the Mass Entropy Transfer field but of Fluid Mechanics.

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